

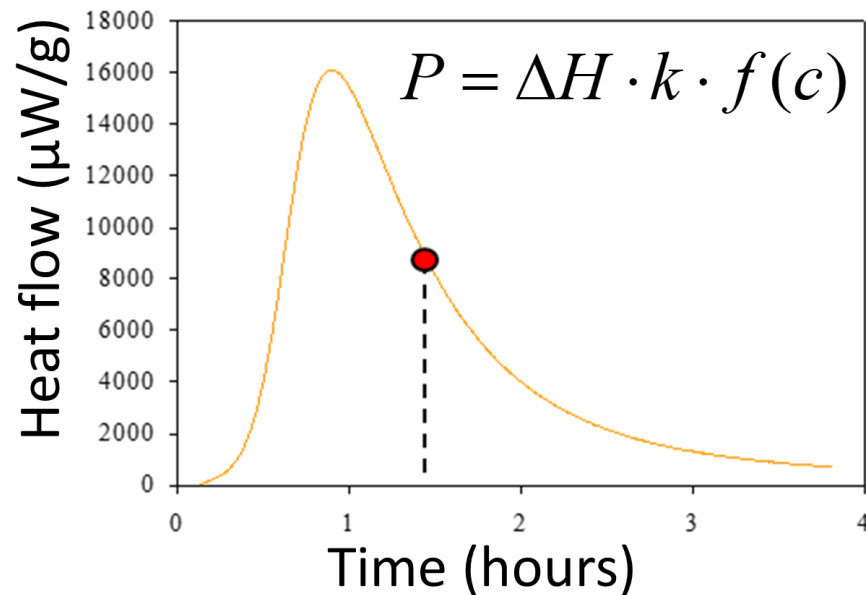
Overview of Microcalorimetry for Material Characterisation – Looking at Battery Behaviour.



Malin Suurkuusk, Ph.D.
TA Instruments
Sweden

An isothermal microcalorimeter measures heat produced or consumed during a physical process or a chemical reaction in terms of heat flow at a constant temperature with microwatt sensitivity or better.

The heat production rate is directly related to the rate of reaction



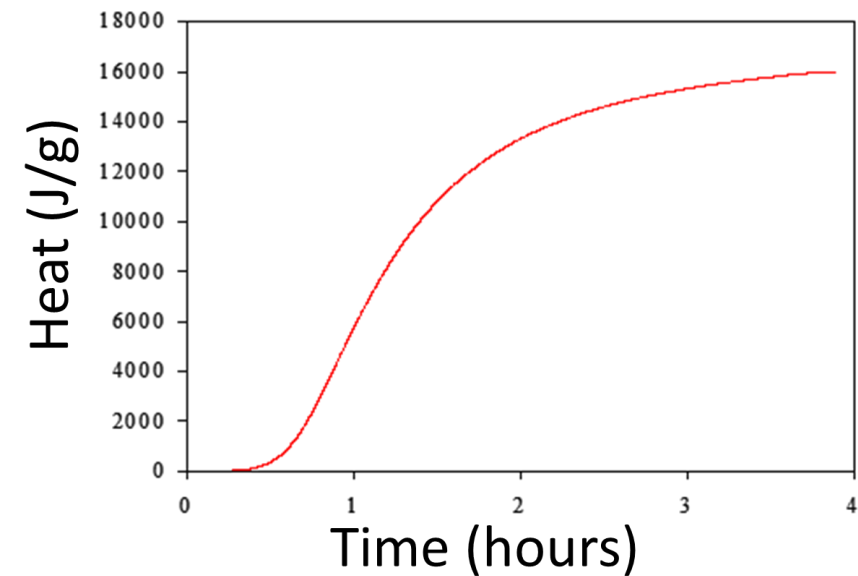
Integration



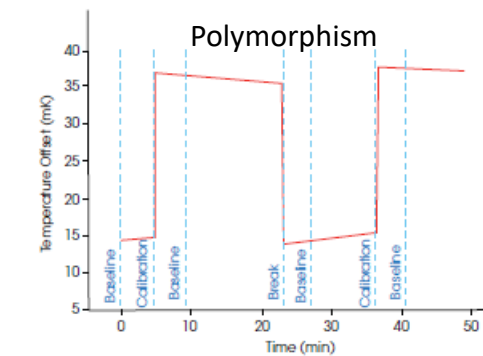
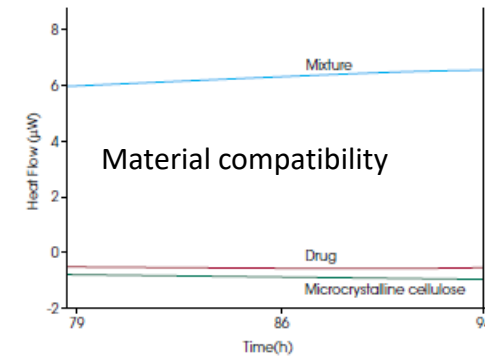
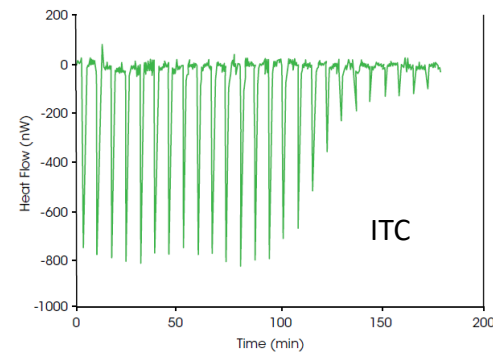
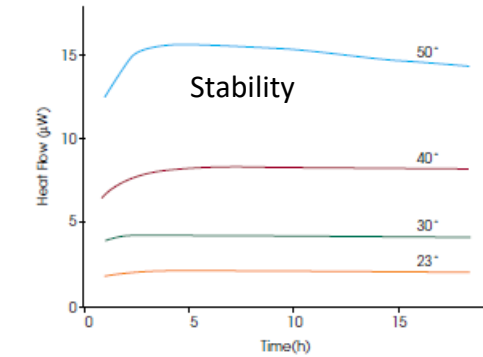
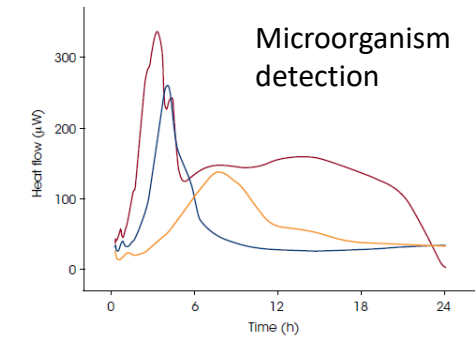
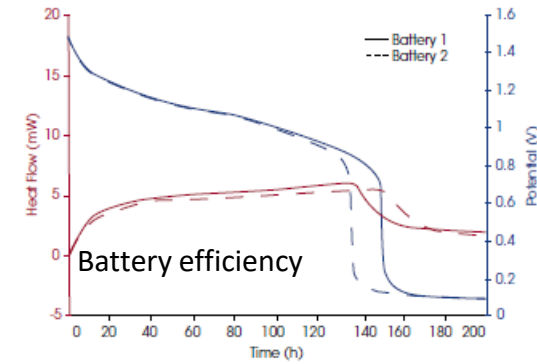
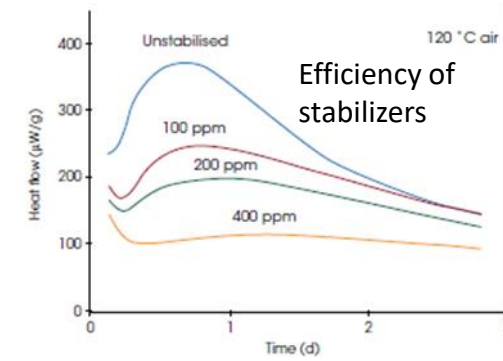
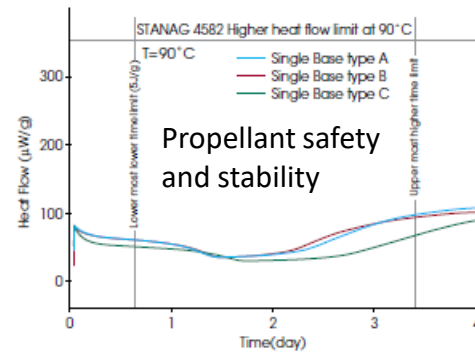
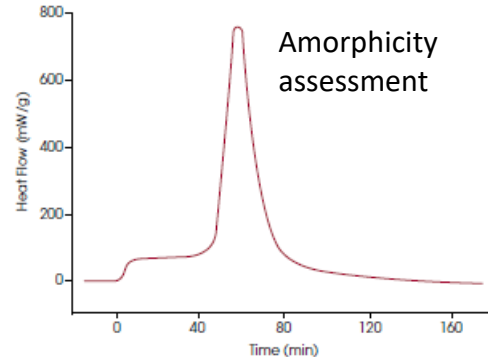
Differentiation



Total heat produced is related to the extent of the reaction



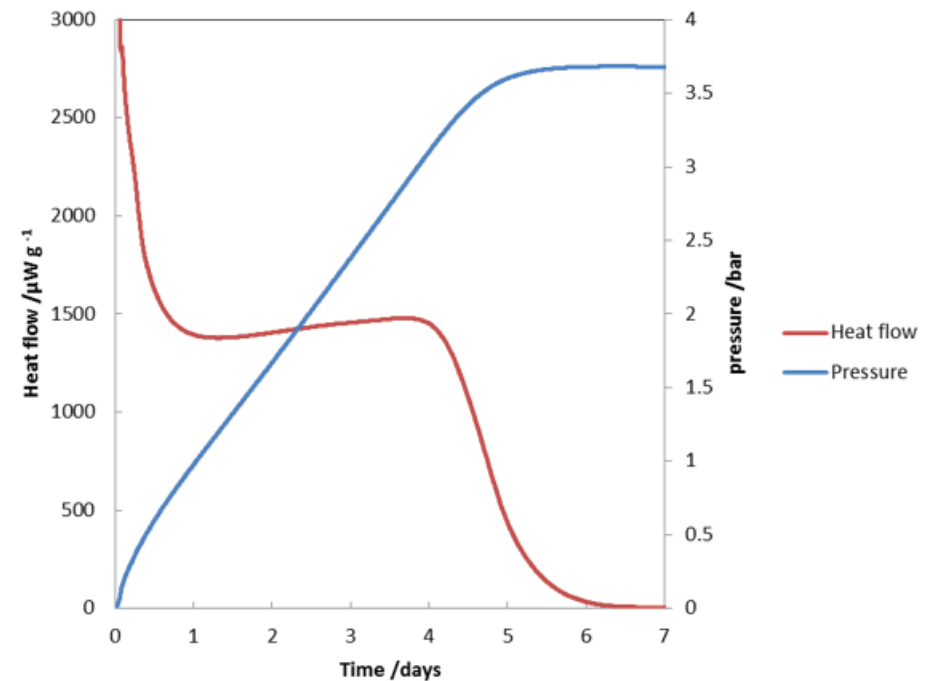
Isothermal calorimetry – a universal technique



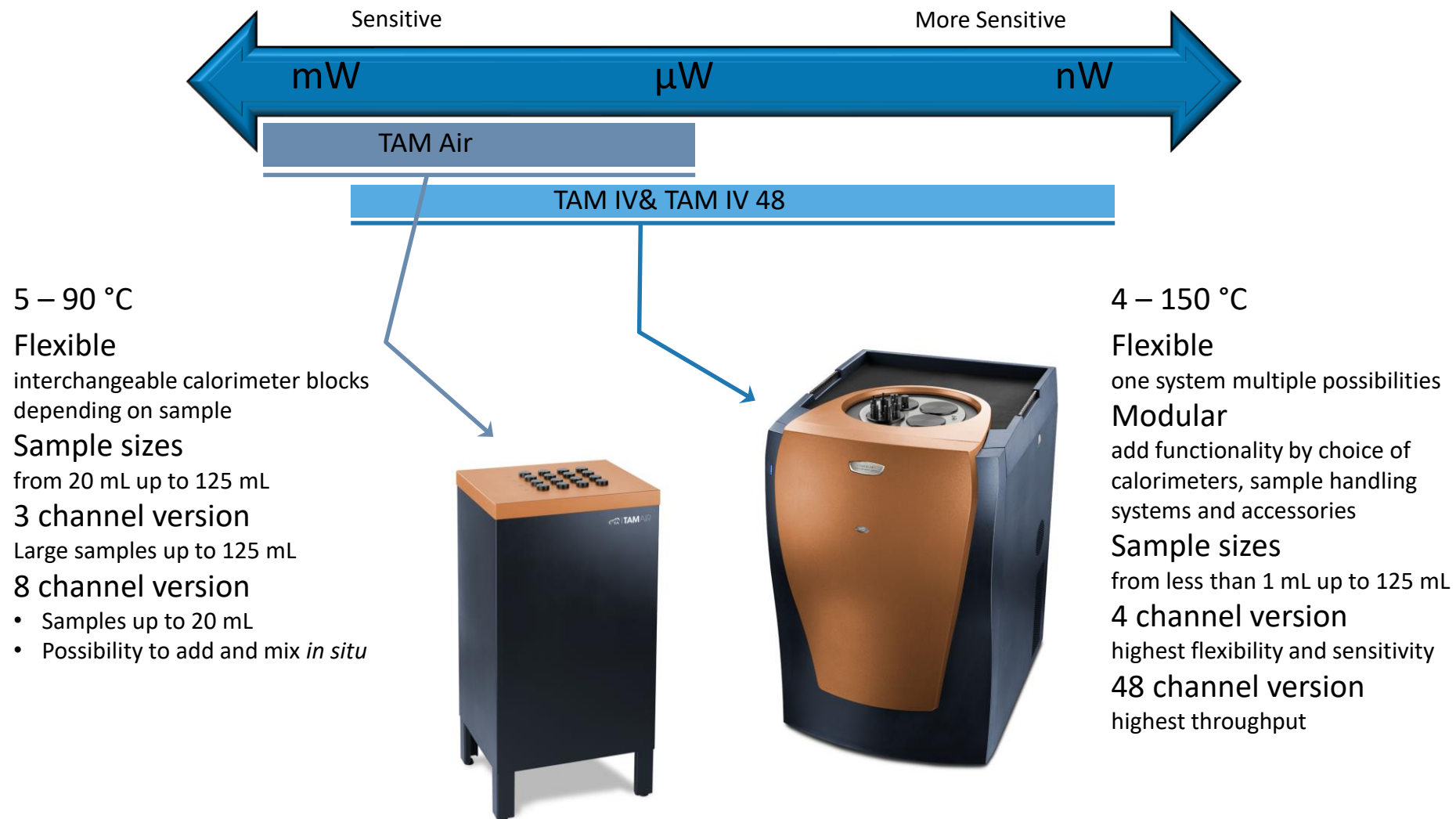
Isothermal calorimetry in material science

Applications include:

- Stability
- Compatibility
- Curing
- Reaction kinetics
- Safety assessments
- Hydration and swelling
- Absorption
- Batteries
- Gas producing reactions



TAM – Thermal Activity Monitor



TAM IV – a flexible multichannel microcalorimetric system

- 1 – 48 individual & independent calorimeters
- nW - mW sensitivity depending on calorimeter
- <1 ml to 20 ml and 125 ml sample sizes
- Sample handling system with possibilities to
 - Add and mix
 - Perfuse gas or liquid over a sample
 - Control and continuously or stepwise change RH or gas/liquid composition
 - Measure pressure changes
 - Dissolve solid samples
 - Add additional probes for increased specificity
- One system – multiple possibilities



TAM IV – A Modular Microcalorimeter System

- The microcalorimeter system base is the TAM IV thermostat.
- Functionality and measuring capacity is obtained by adding:
 - Calorimeters
 - Sample handling systems (ampoules)
 - Auxiliary equipment



Why use TAM for battery testing?

- Only method that directly measures the occurrence of non-current producing reactions under load
- Very sensitive method to assess self-discharge, sometimes the only method
- Non-destructive
- Simple
- Continuous



Measured battery power

Battery under load

$$\frac{dq}{dt} = \frac{E^2}{R_{\text{resistor}}} + \frac{E^2}{R_{\text{battery}}} + \textit{parasitic reactions}$$

Self-discharging battery

$$\frac{dq}{dt} = \frac{E^2}{R_{\text{battery}}} + \textit{parasitic reactions}$$

Parasitic reactions are material compatibility issues,
side reactions not part of the battery chemistry

Battery applications

- Stability and shelf life
 - Self discharge will show heat production in an open circuit
- Life cycle
 - Variance of thermal performance during charge/discharge cycles
 - Evaluation of number of charge cycles possible
- Efficiency
 - How much of the total power generated is available for the external device?
 - Internal resistance
 - Depends on temperature, current draw, age, charge cycles etc.

Battery applications cont.d

- Performance
 - Depends on the discharge conditions such as the magnitude of the current, the allowable terminal voltage of the battery, temperature and other factors
- Quality control
 - For critical applications, e.g. pacemaker batteries, to make sure there are no unwanted processes and internal short circuits
- Development
 - New type of electrochemical components
 - Compatibility
 - Evaluation of individual battery components

Battery tests can be performed as function of

- Temperature
- Size & shape of battery
- Chemistry
- Current draw
- Age
- State of charge



Calorimeters for battery testing

➤ Macrocalorimeter

- Volume: 210 mL
- Sample size up to 61.2 mm diameter and 72 mm high

➤ 20 ml Micro or Mini/Multicalorimeter

- For smaller batteries and where higher sensitivity is needed also the 20 ml calorimeters can be used.

➤ Battery measurements accessories

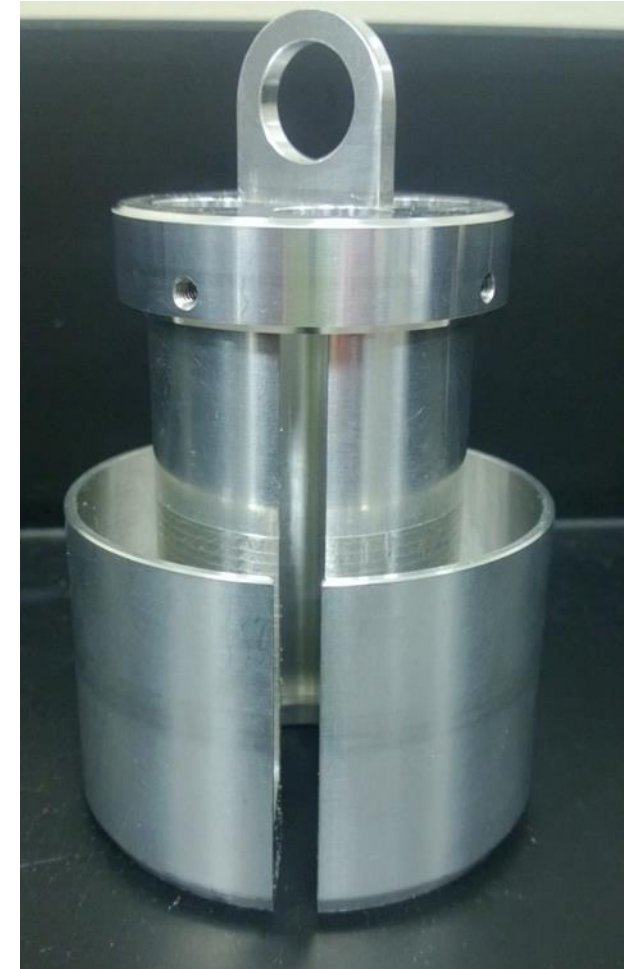
- Insert for standard batteries
- Lifters with wiring capabilities



Battery fixtures

Purpose of fixture

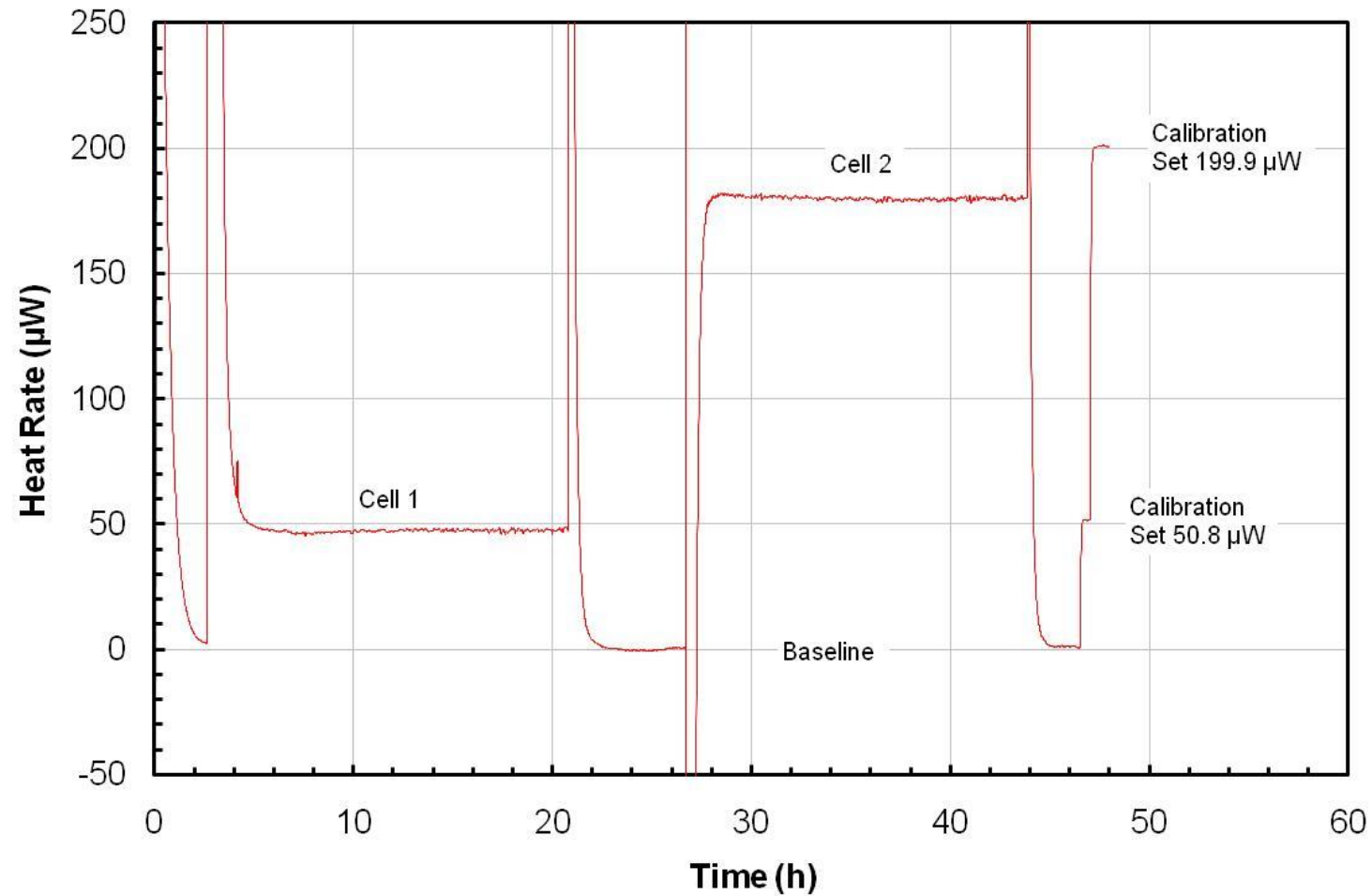
- To optimise the thermal contact between the battery and the heat flow detector
- To make sure the position of the battery in the calorimeter is reproducible
- To facilitate the insertion of the battery into the calorimeter
- To avoid short-circuiting the battery and facilitate connections to the battery for load measurements
- Available for C- and D- cell batteries as well as 18650 batteries



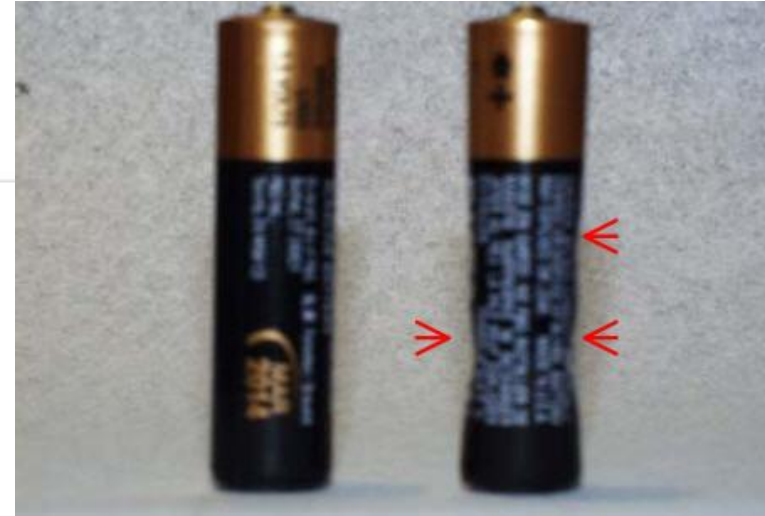
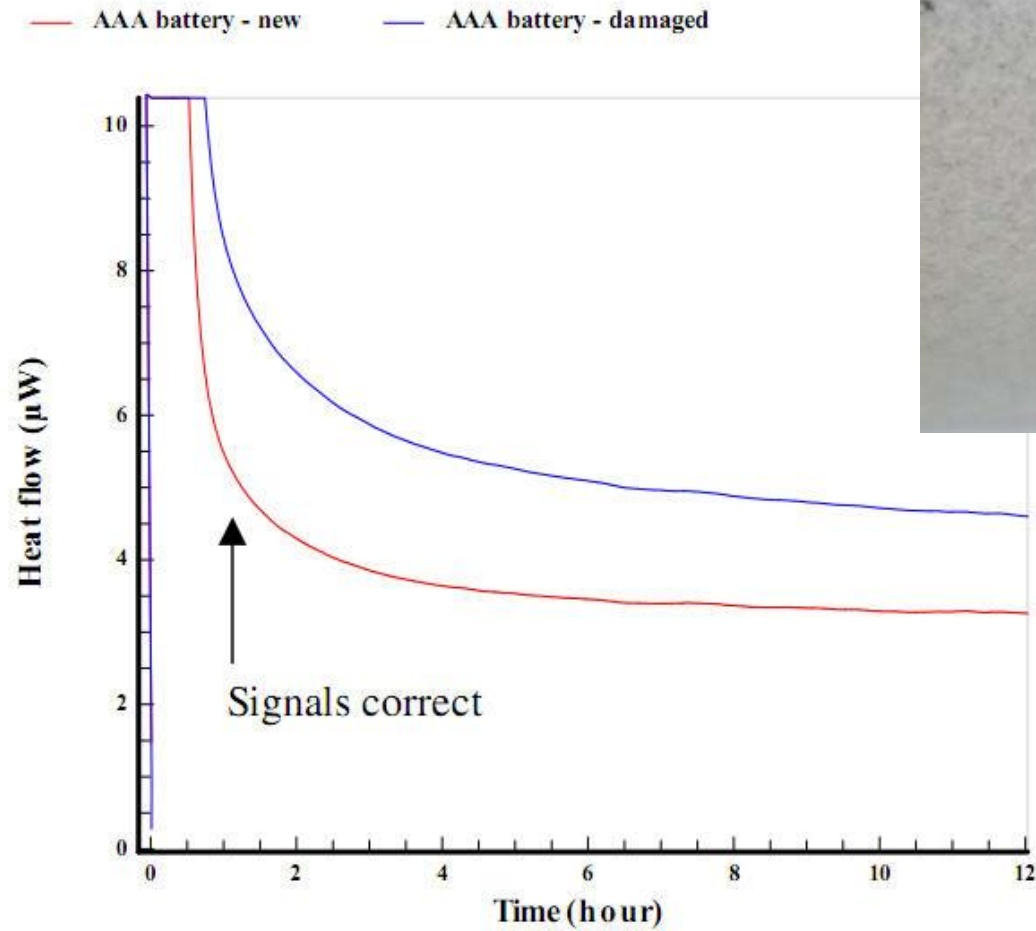
Open circuit discharge



Example on two different batteries and calibrations



Damaged batteries



Tests for short-circuited batteries

- Internally short-circuited battery

	Open circuit voltage (V)	Load voltage (V)	Heat output (μW)
Good cell	2.8	2.75	25
Shorted cell	< 2.5	< 2.0	150 - 200

- Pacemaker batteries

- Tested with a well defined circuit in the calorimeter

Good batteries < 20 μW

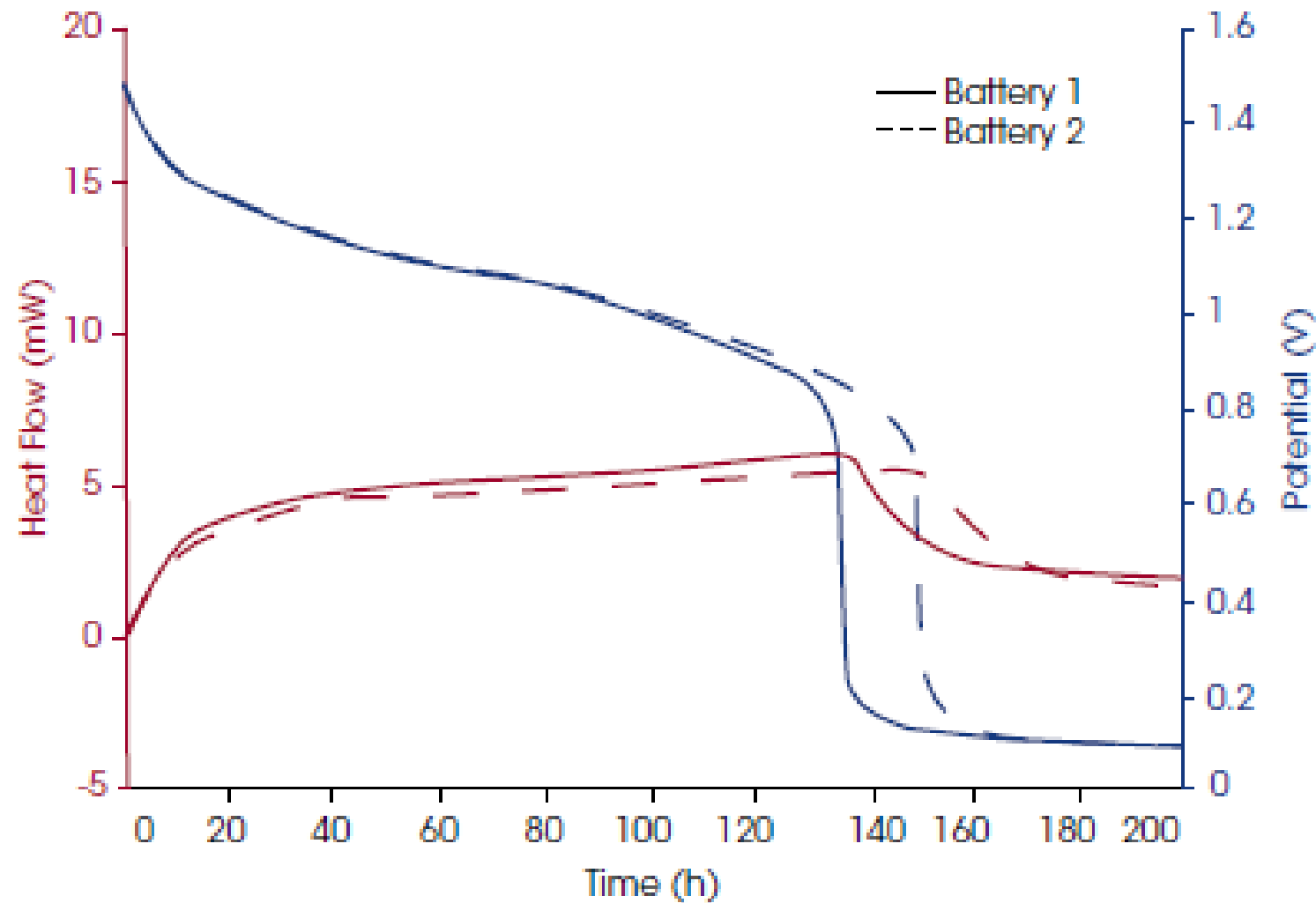
Bad batteries > 70 μW



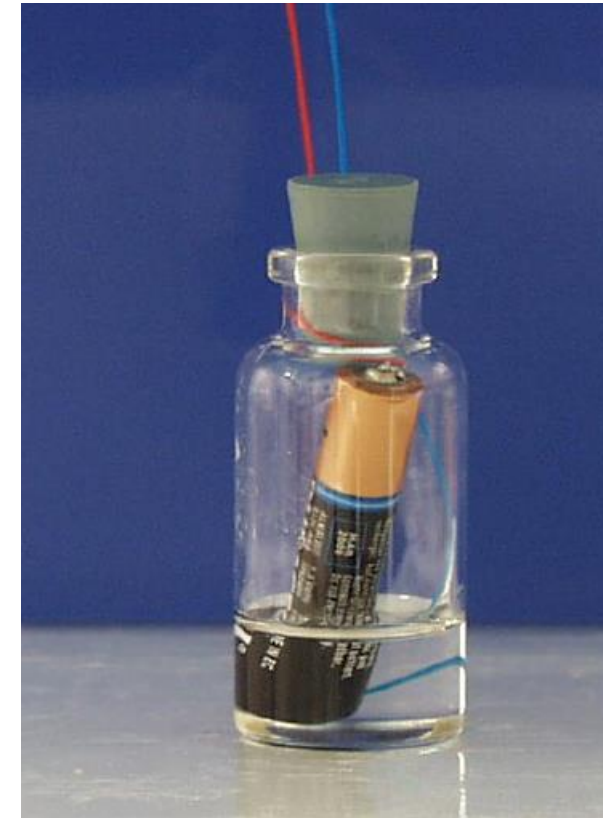
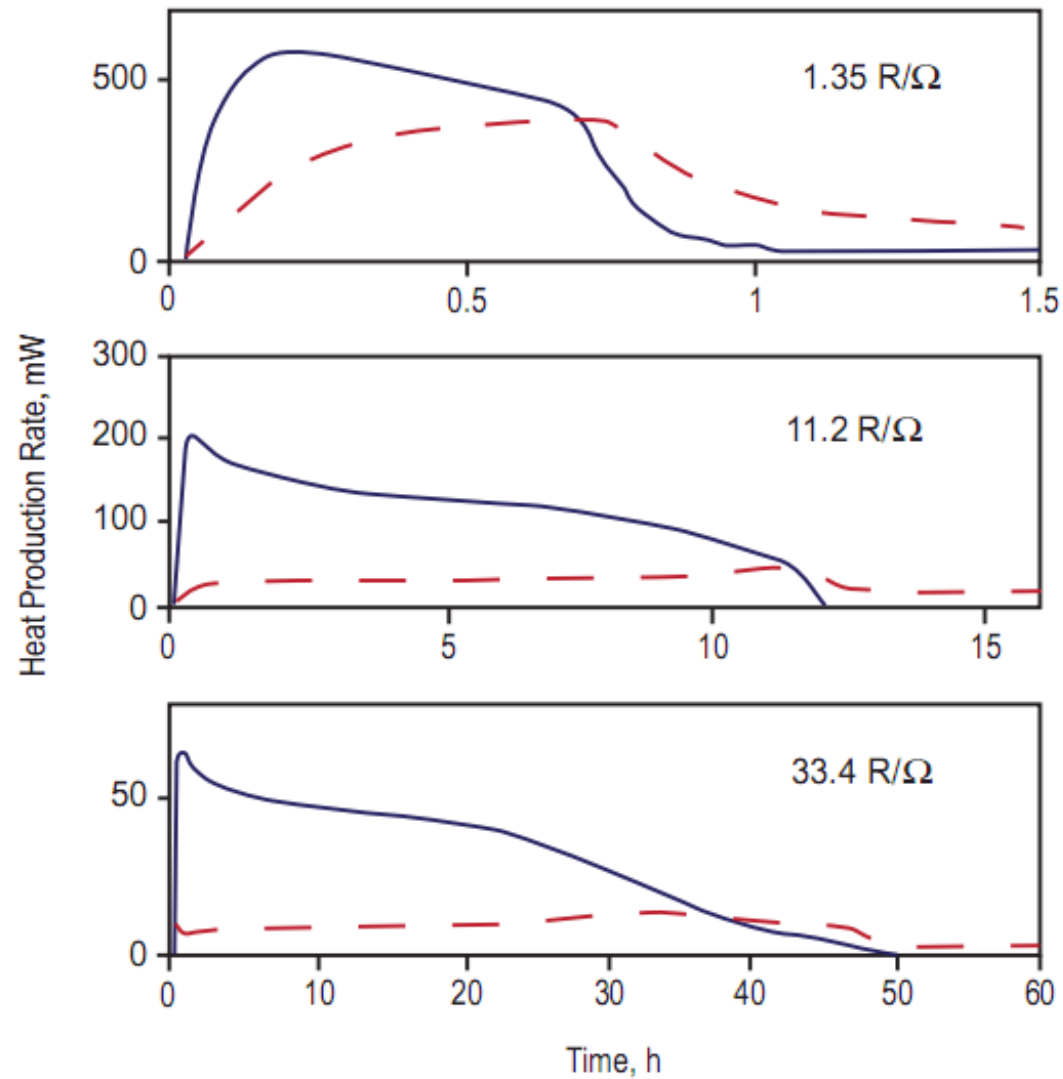
Charge - Discharge



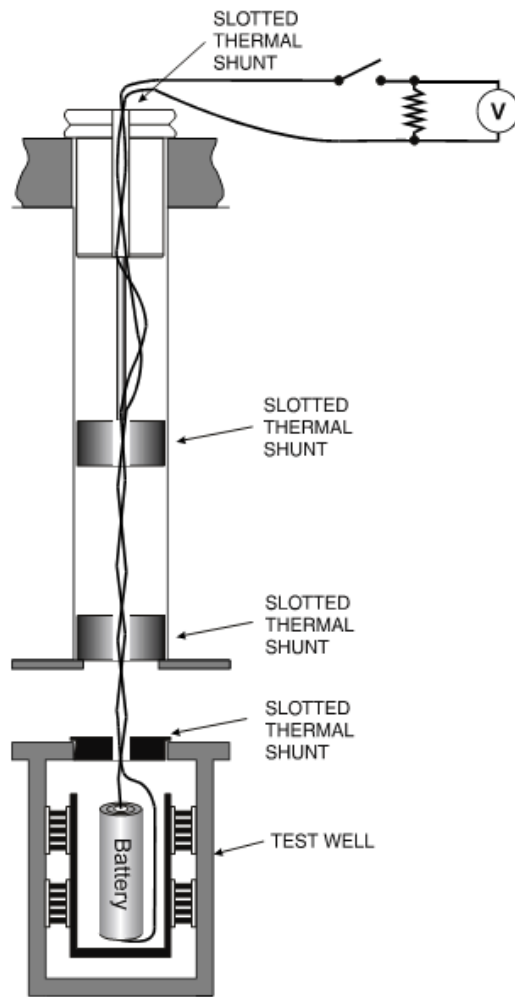
Heat flow and voltage measurements



Battery under load



Battery with load in the calorimeter



Experimental setup

A set 18650 batteries had metal tabs spot-welded to each terminal of each battery.

Two wires were soldered to each tab – a black set to the negative side and a red set to the positive side.

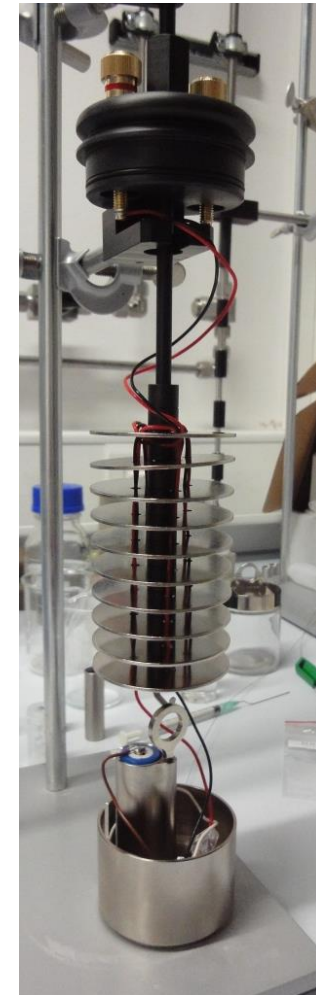
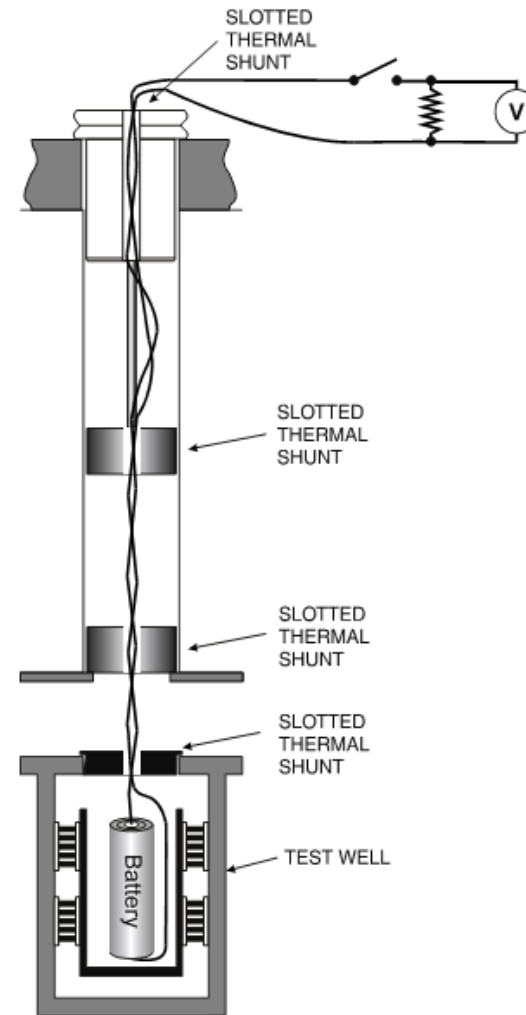
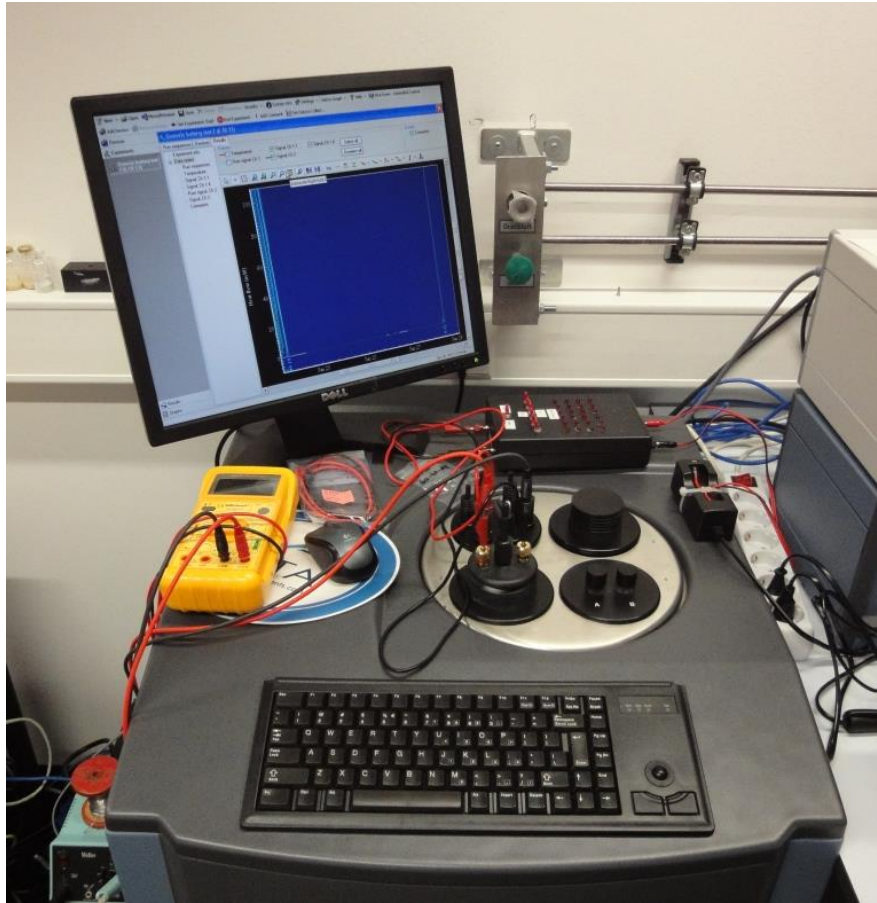
The ends were then covered with capton tape to electrically insulate the terminals

The batteries were placed inside adapters and the wires through the holes in the shunts, then the adapters were placed inside the TAM.

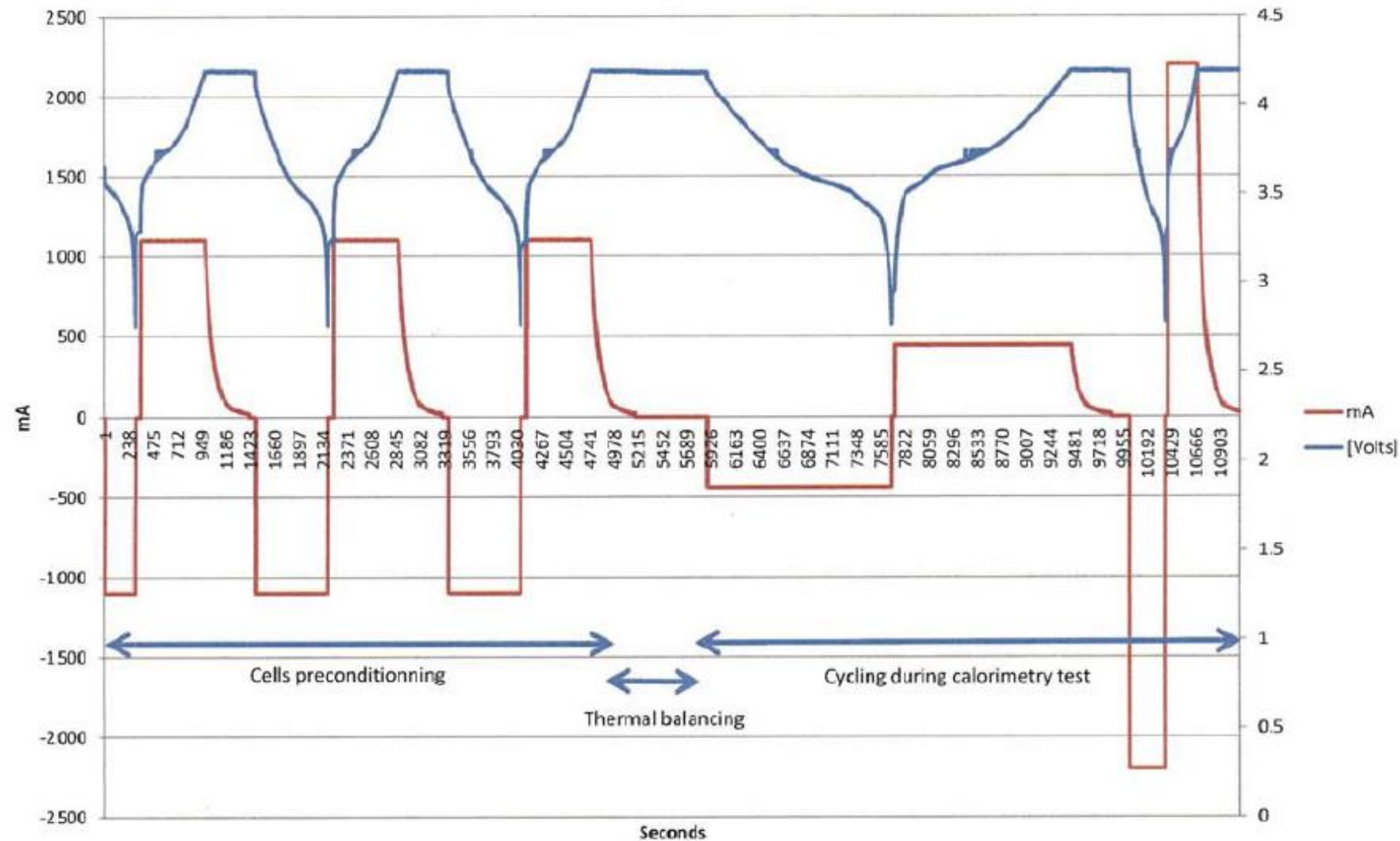
A battery tester was connected to one set of wires, while the other set were connected to a voltage measurement device.



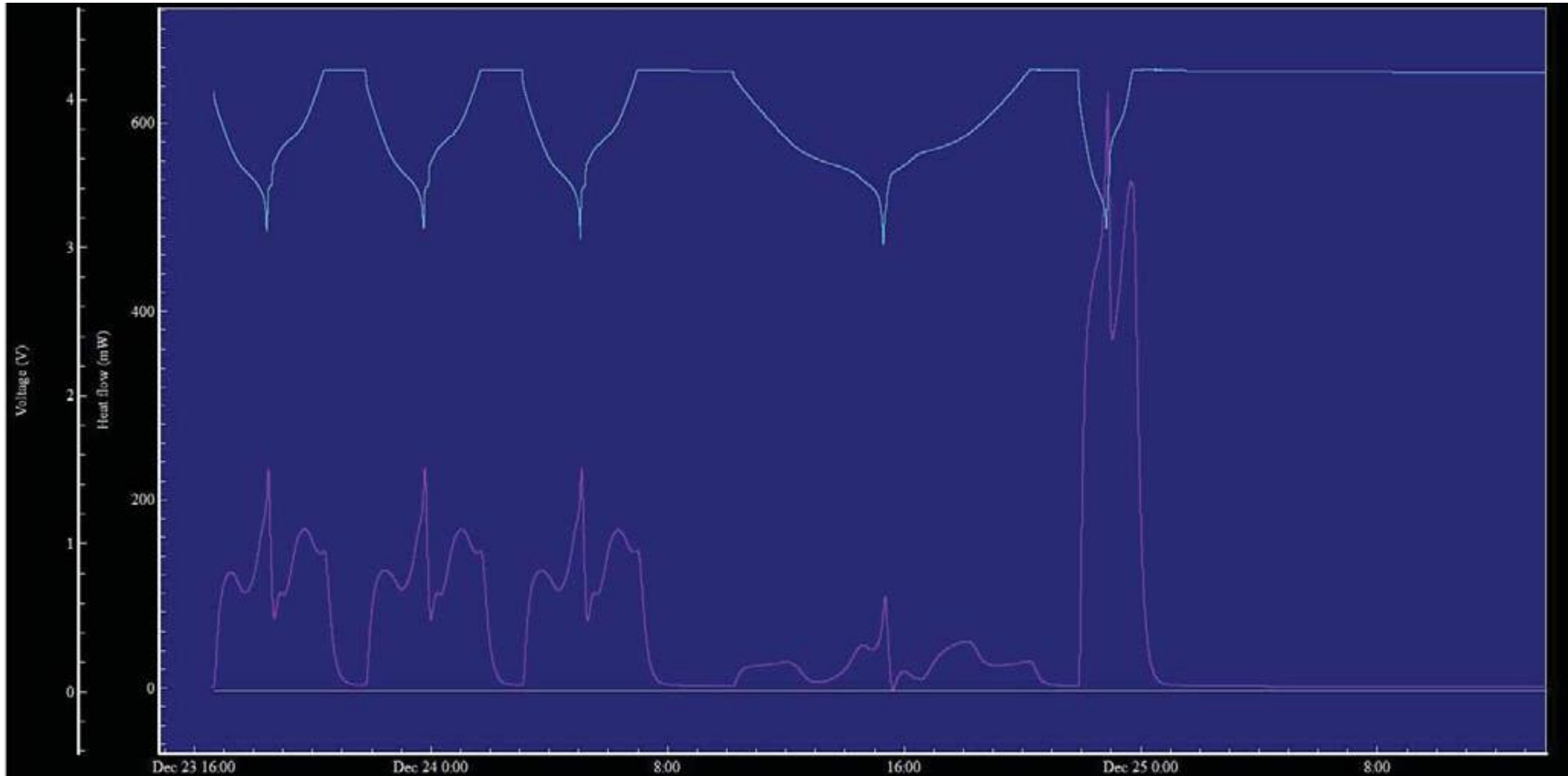
Battery under load



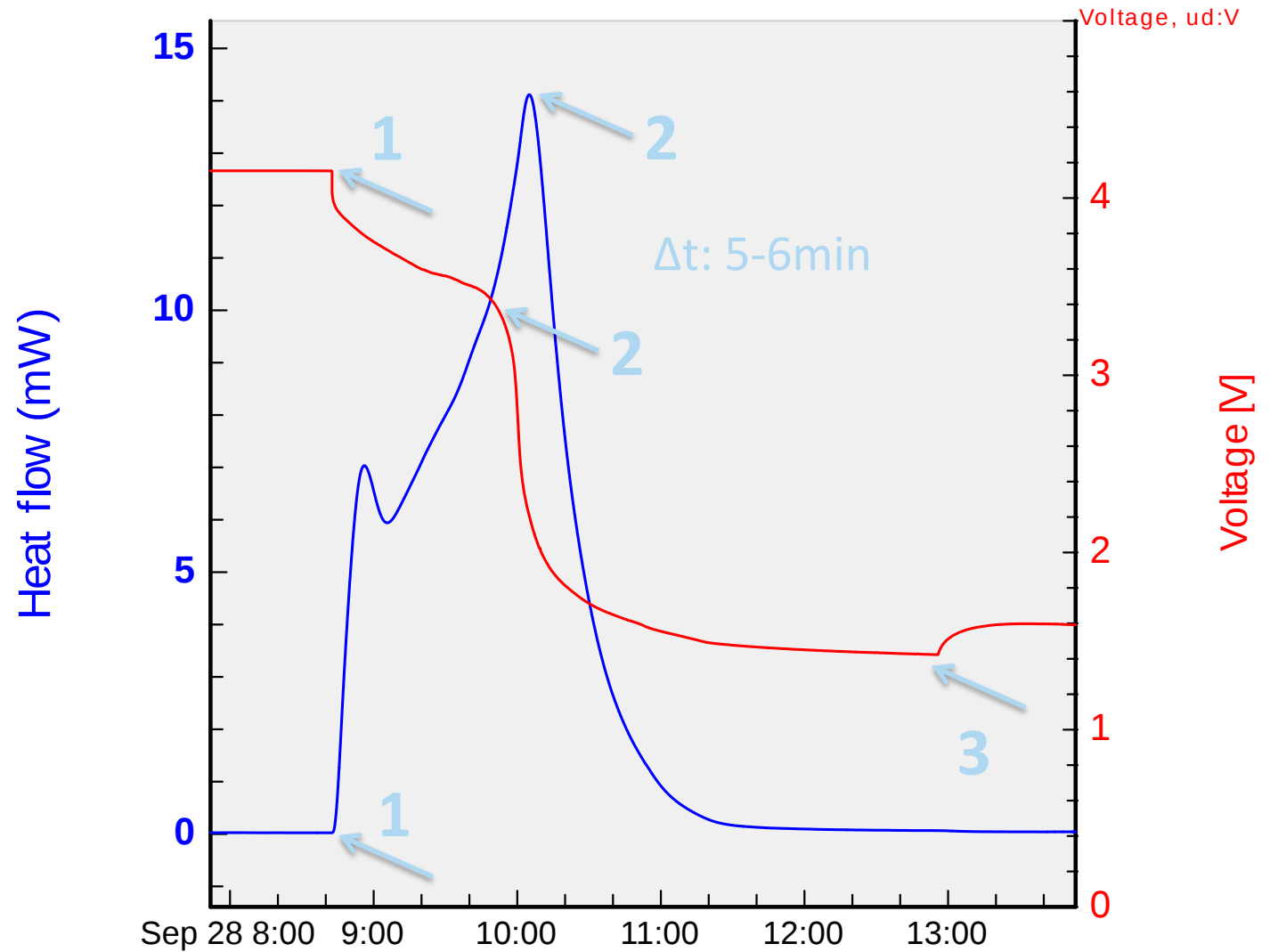
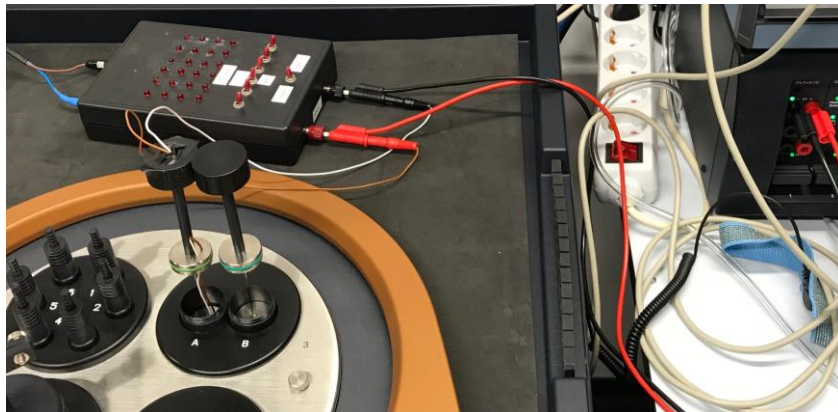
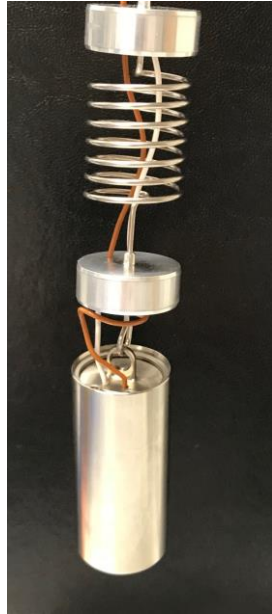
Experimental program



Results



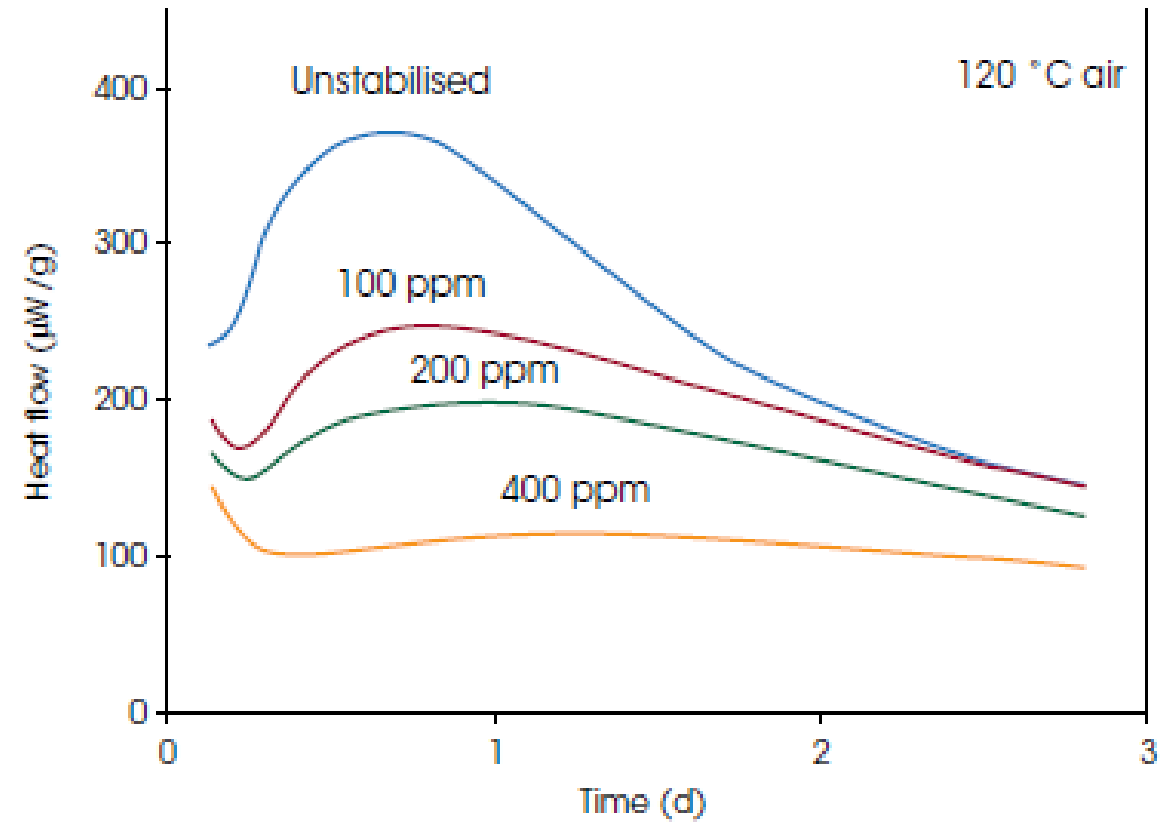
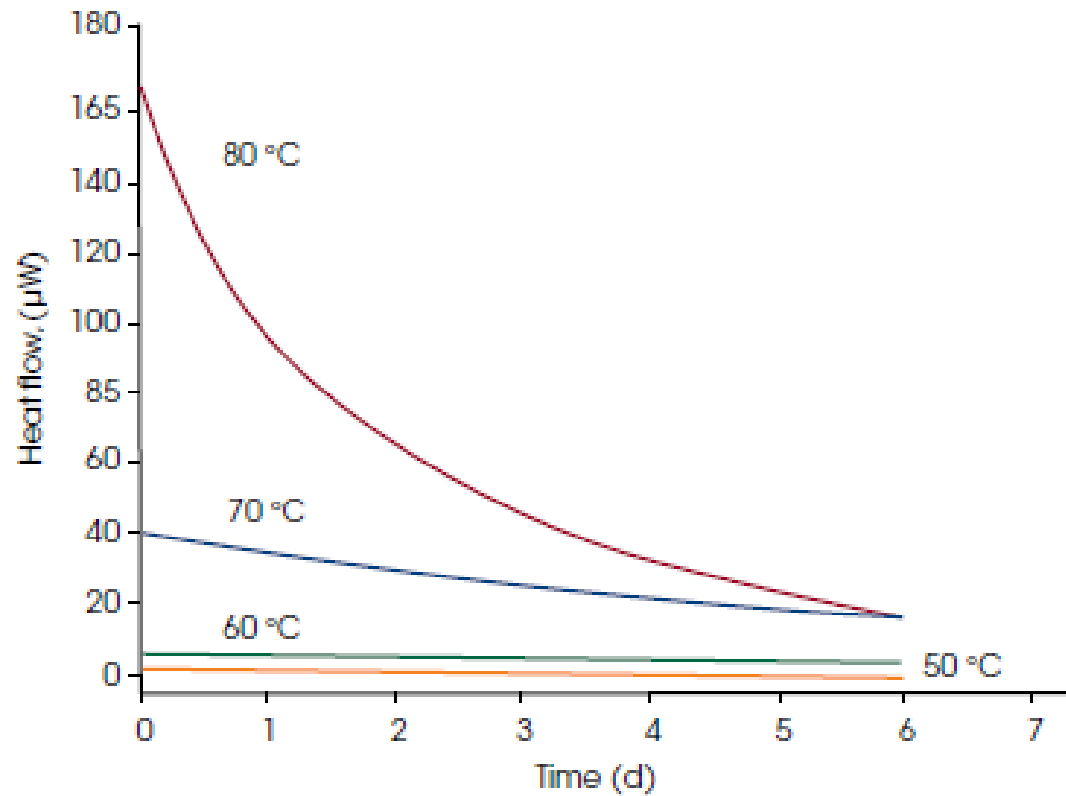
Button cell battery- Li R2032 - Discharge



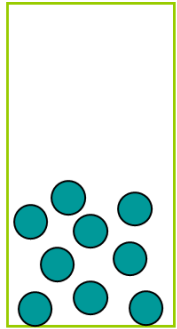
Stability and Compatibility



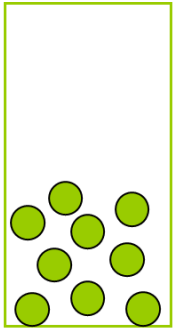
Battery individual component stability measurements



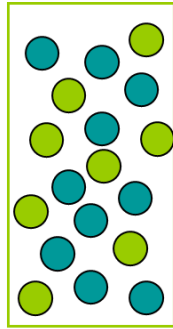
Battery component compatibility testing



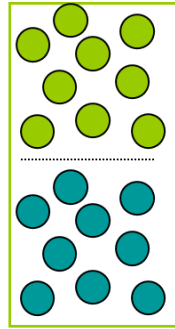
Sample A



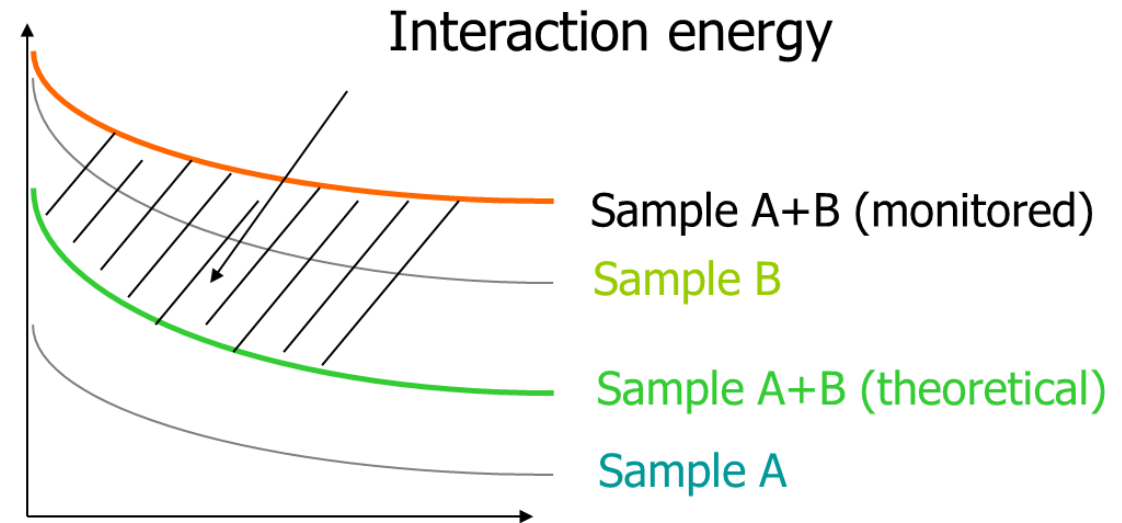
Sample B



Sample A+B

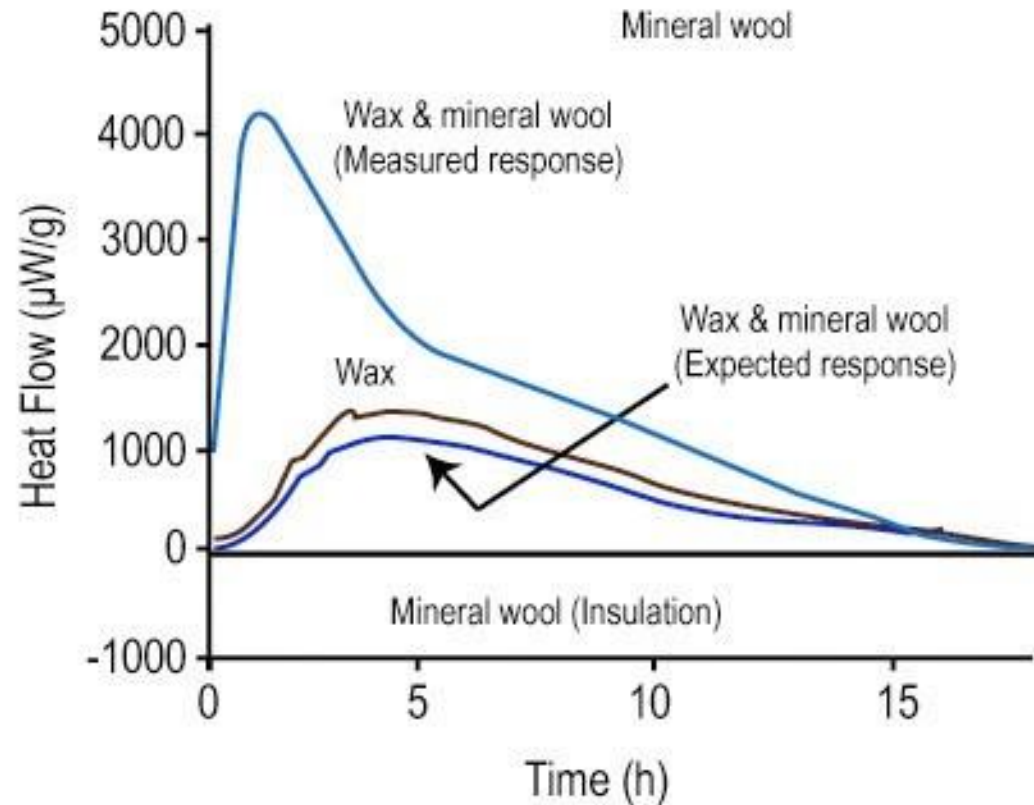


Sample A+B



Theoretical curve $\neq$ Monitored curve $\longrightarrow$ Not compatible !

Compatibility Between Wax and Mineral Wool



Data provided by Svensson, *Bodycote Materials AB*, Sweden (2003)

Summary of using TAM for battery testing

- Only method that directly measures the occurrence of non-current producing reactions under load
- Very sensitive method to assess self-discharge, sometimes the only method
- Non-destructive
- Simple
- Continuous

In addition to whole device testing individual component stability and compatibility can be evaluated





Thanks!